

Post - *simple cubic* structures in compressed phosphorus and calcium: electronic origin

Valentina Degtyareva



*Institute of Solid State Physics,
Russian Academy of Sciences
Chernogolovka, Russia*

Outline

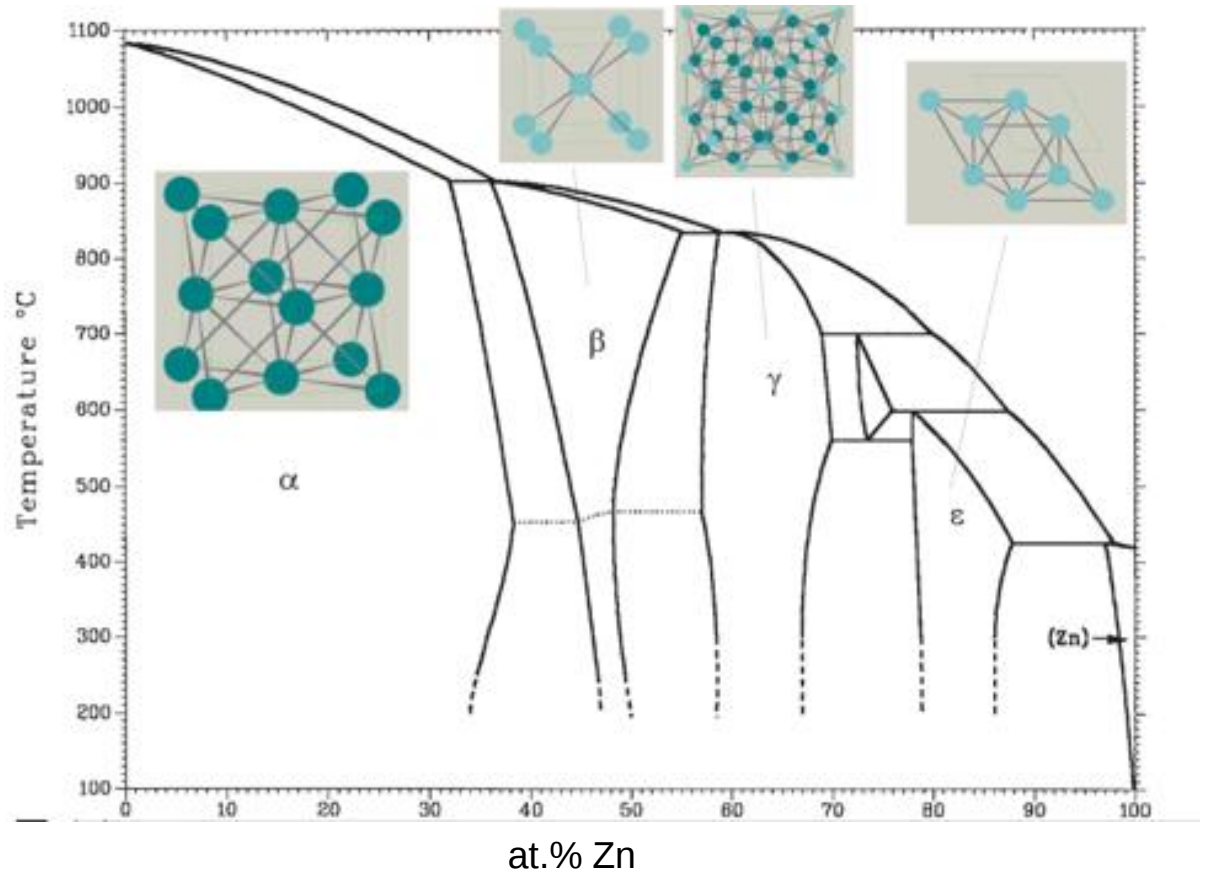
- Main factors of crystal structure stability
- Concept of the Fermi Sphere - Brillouin Zone interaction: Cu-Zn alloy system
- An incommensurately modulated phase P-IV: consideration with a commensurate approximant
- *Simple cubic* phase in Ca and its distortion on further compression: *pressure ionization* of core electrons

Phase diagram of the Cu-Zn system



The Age of Bronze

A. Rodin

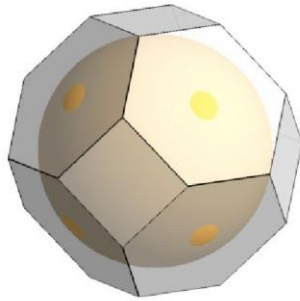


α (fcc) $\rightarrow$ β (bcc) $\rightarrow$ γ (complex cubic) $\rightarrow$ ϵ (hcp)
1.35 $\rightarrow$ 1.5 $\rightarrow$ 1.62 $\rightarrow$ 1.75
(electron / atom)

after Massalsky (1996)

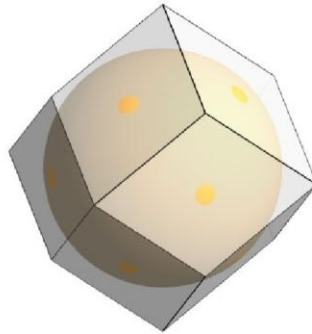
Hume-Rothery phases: Fermi sphere – Brillouin zone interaction

α (fcc) - phase



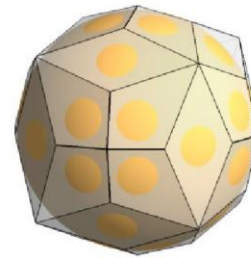
$\{111\}$
 $\{200\}$

β (bcc) - phase



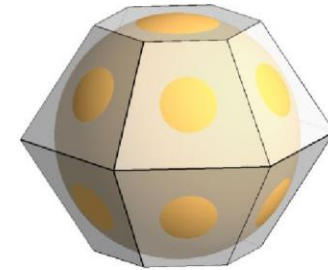
$\{110\}$

γ - complex cubic



$\{411\}$
 $\{330\}$

ϵ (hcp) - phase



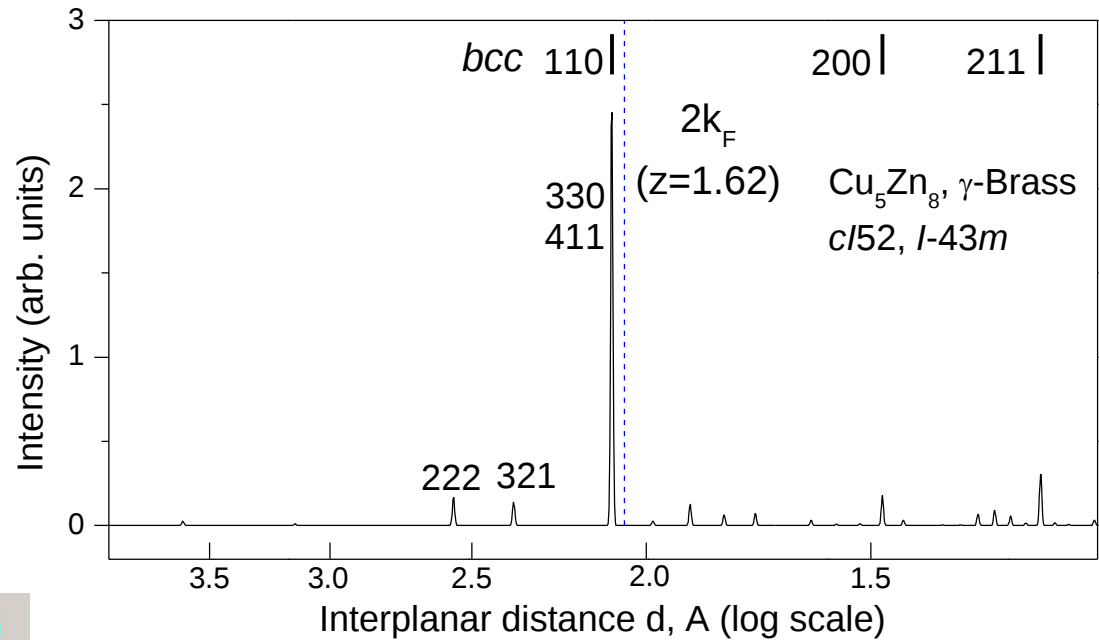
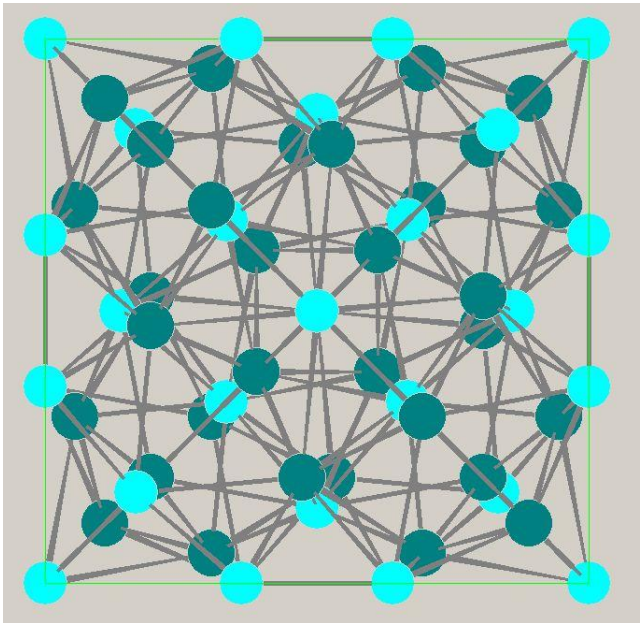
$\{002\}$
 $\{101\}$

Fermi sphere – energy surface of free valence electrons, radius $k_F = \left(\frac{3\pi^2 z}{V} \right)^{1/3}$

Brillouin zone – planes in reciprocal space with vector $q_{hkl} = \frac{2\pi}{d_{hkl}}$

Interaction (condition of phase stability): $k_F \approx \frac{1}{2} q_{hkl}$

The γ -phase Cu_5Zn_8 :
 complex cubic structure
 52 atoms per unit cell,
 space group $I\bar{4}3m$,
 lattice parameter
 $a = 8.86 \text{ \AA}$
 [Pearson 1976].



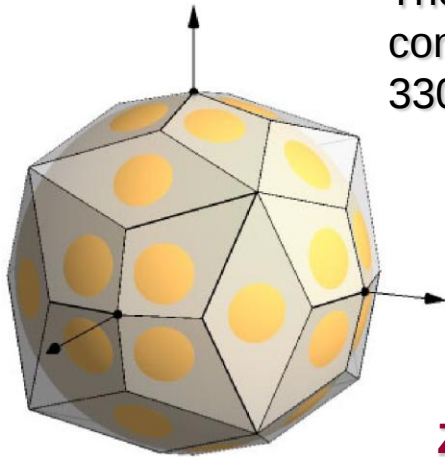
$3 \times 3 \times 3$ supercell of *bcc*

2 out of 54 atoms are removed
 the remaining 52 atoms are
 slightly displaced

so that an additional
 reflection $\{411\}$ appears.

Hume-Rothery effects stabilize low-symmetry structures

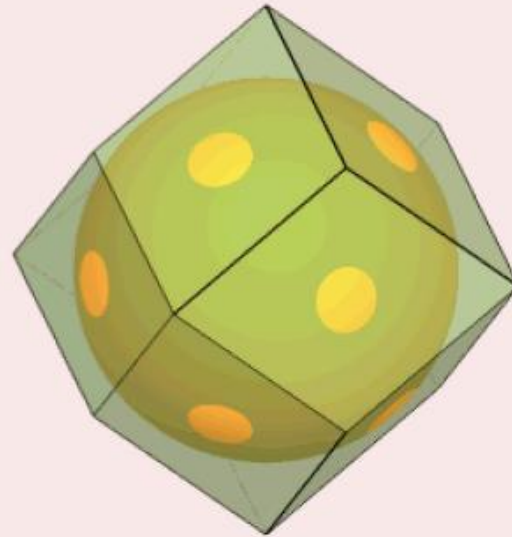
γ - phase Cu_5Zn_8



The Brillouin zone
consists of 36 planes
330 and 411 types

**Zone filling
by valence electrons
is ~93%**

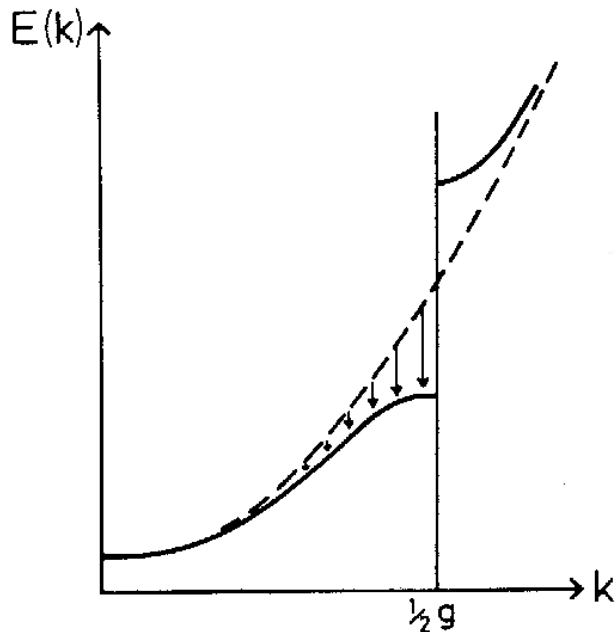
“Jones barrel” is filled



Discussion of stability of the γ -brass

The criterion of stability for the crystal structure of Hume-Rothery phases is a contact of Brillouin zone planes to the Fermi sphere.

Formation of an energy gap at the Brillouin zone boundary lowers the kinetic energy of the free electrons and accounts for the stability of the crystal structure.



Band structure energy E_{BS}

$$E = E_o + E_{Ewald} + E_{BS}$$
$$E_{Ewald} = -\alpha \frac{(Ze)^2}{2r_0} \quad E_{BS} = \sum_q |S(q)|^2 \Phi(q)$$

Volume scaling:

$$\sim V^{-1/3}$$

$$\sim V^{-2/3}$$

Enhancement of the Hume-Rothery arguments at compression

BRIZ – a program for the FS-BZ visualization

Smirnova I.S. & Degtyareva V.F.

Z.Kristallog. **222**, 718. 2007



Aperiodic structures in elements

1. Incommensurate host-guest structures (1999-2008)
2. Incommensurately modulated structures (2003-2007)
3. Long-period superstructures (2001-2009)

IA	IIA	IIIB	IVB	VB	VIB	VII B	VIII B					IB	IIB	IIIA	IVA	VA	VIA	VIIA	VIIIA
1 H																		2 He	
3 Li	4 Be											5 B	6 C	7 N	8 O	9 F	10 Ne		
11 Na	12 Mg											13 Al	14 Si	15 P	16 S	17 Cl	18 Ar		
19 K	20 Ca	21 Sc	22 Ti	23 V	24 Cr	25 Mn	26 Fe	27 Co	28 Ni	29 Cu	30 Zn	31 Ga	32 Ge	33 As	34 Se	35 Br	36 Kr		
37 Rb	38 Sr	39 Y	40 Zr	41 Nb	42 Mo	43 Tc	44 Ru	45 Rh	46 Pd	47 Ag	48 Cd	49 In	50 Sn	51 Sb	52 Te	53 I	54 Xe		
55 Cs	56 Ba	57 La	72 Hf	73 Ta	74 W	75 Re	76 Os	77 Ir	78 Pt	79 Au	80 Hg	81 Tl	82 Pb	83 Bi	84 Po	85 At	86 Rn		
87 Fr	88 Ra	89 Ac																	
			58 Ce	59 Pr	60 Nd	61 Pm	62 Sm	63 Eu	64 Gd	65 Tb	66 Dy	67 Ho	68 Er	69 Tm	70 Yb	71 Lu			
			90 Th	91 Pa	92 U	93 Np	94 Pu	95 Am	96 Cm	97 Bk	98 Cf	99 Es	100 Fm	101 Md	102 No	103 Lr			

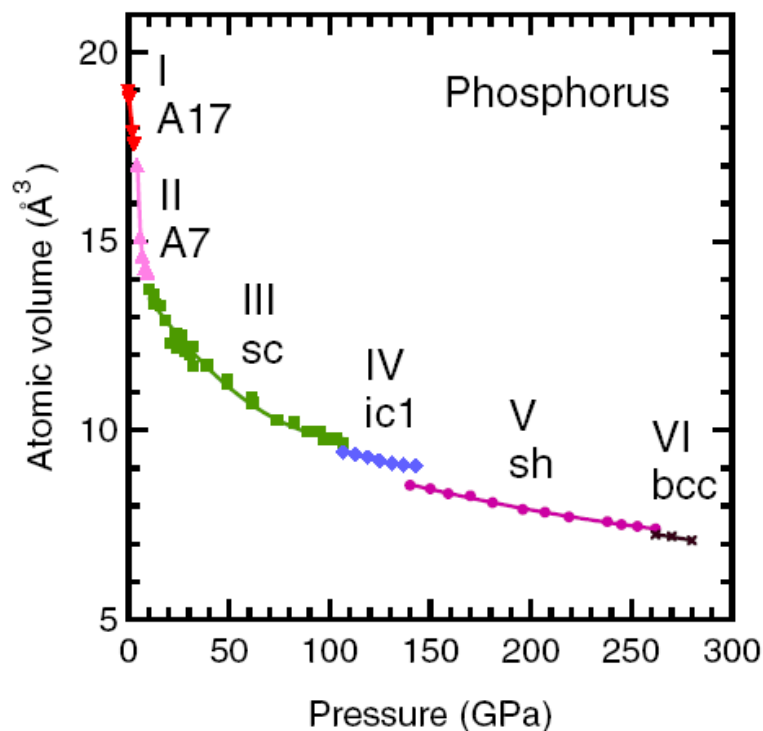
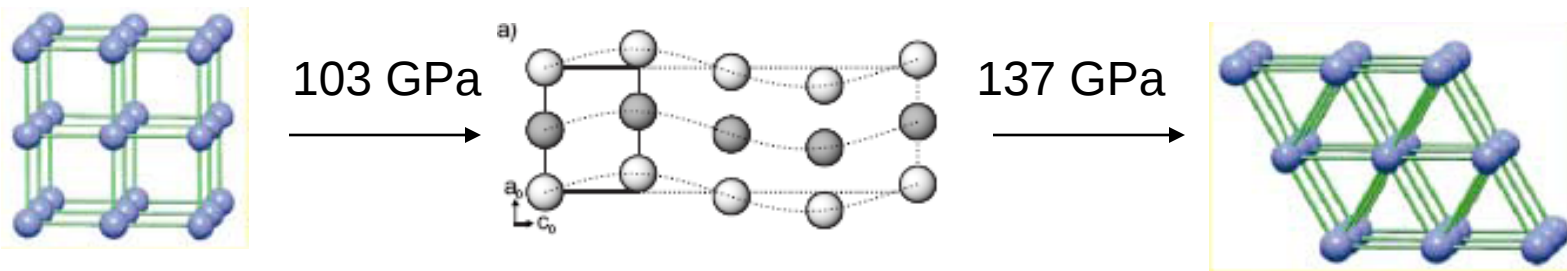
Elemental Complexity ↔ Elemental Simplicity

Summary of sc – cP1

Element (group)	Conditions	Refs
Po (VI)	Ambient pressure	Pauling File
P (V)	High pressure (10-103 GPa)	J. C. Jamieson, <i>Science</i> 139 , 129 (1963) Y. Akahama, M. Kobayashi and H.Kawamura, <i>Phys. Rev. B</i> 59 8520 (1999)
As (V)	High pressure (25-48 GPa)	T. Kikegawa and H. Iwasaki, <i>J. Phys. Soc. Jpn.</i> 56 , 3417 (1987)
Ca (II)	High pressure (32-113 GPa)	H. Olijnyk and W. B. Holzapfel, <i>Phys. Lett.</i> 100A , 191 (1984) T. Yabuuchi, Y. Nakamoto, K. Shimizu, and Y. Kikegawa, <i>J. Phys. Soc. Jpn.</i> 74 , 2391 (2005) Q. F. Gu, G. Krauss, Yu.Grin, and W.Steurer, <i>Phys. Rev. B</i> 79 , 134121 (2009)
Se (VI)	After shock compression	V. F. Degtyareva, V.N. Sikorov, <i>Solid State Physics</i> 19 , 2201 (1977)
Se (VI)	Thin film (sublimation in vacuum)	Andrievskii, I. D. Nabitovich, and P. I. Kripyakevich, <i>Soviet Phys.-Doklady</i> 4 , 16 (1959)

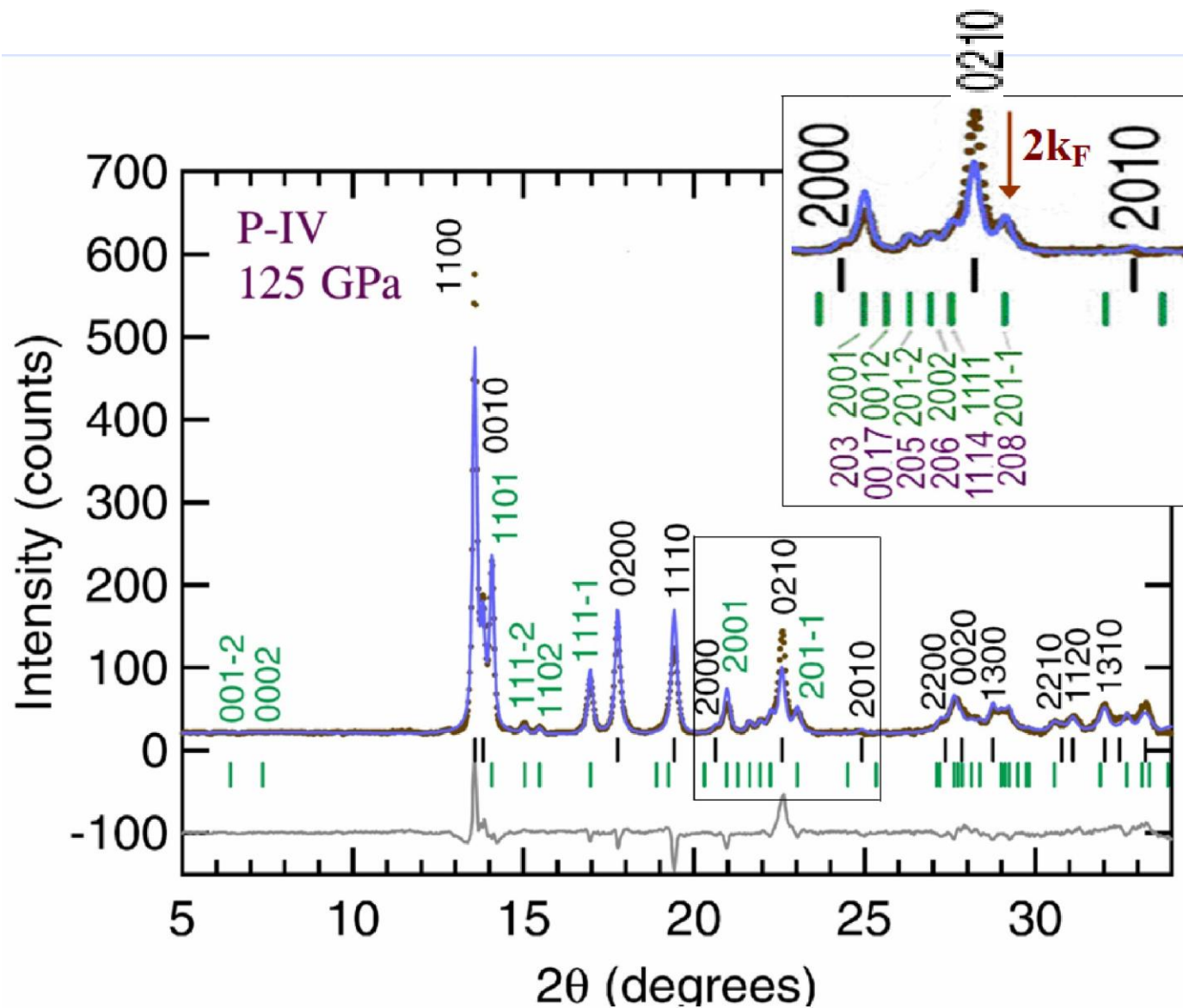
Structural sequence under pressure in P (group-V element)

P	$\text{orth} \xrightarrow{5} \text{hR2} \xrightarrow{10} \text{sc} \xrightarrow{103} \text{inc.mod.} \xrightarrow{137} \text{sh} \xrightarrow{260} \text{bcc} < 280 \text{ GPa}$
----------	--

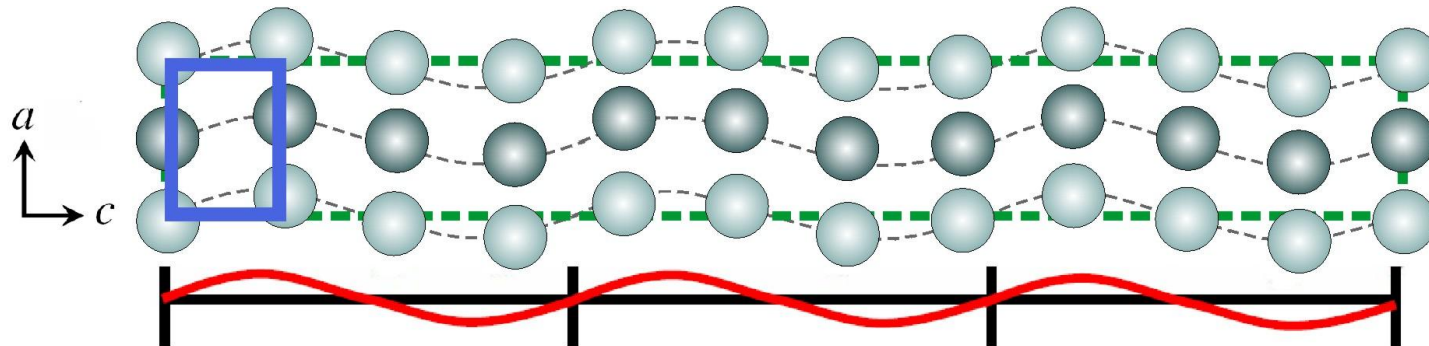


[Fujihisa et al PRL 2007]
[Marques et al PRB 2008]

The oC2 P-IV structure: 5 valence electrons



The oC2 P-IV structure

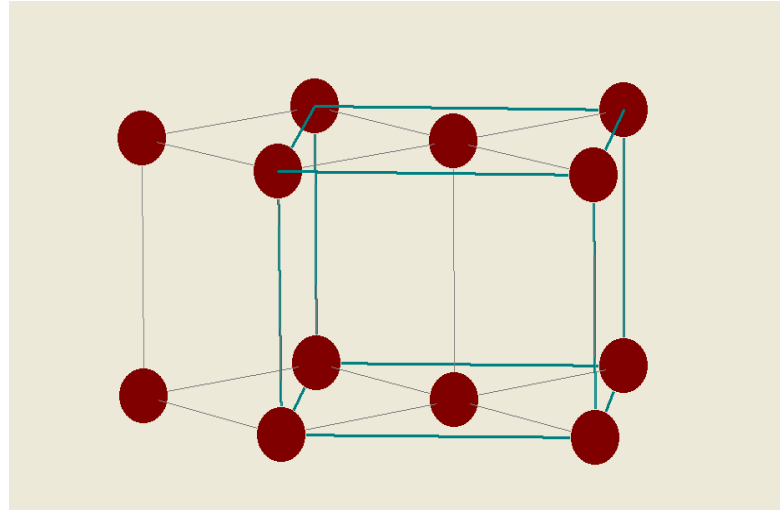


Fujihisa et al, PRL **98**, 175501 (2007)
 $a=2.772$ Å, $b=3.215$ Å, $c=2.063$ Å
 $Cmmm$ (00γ) s00
 125 GPa atomic volume of 9.19 Å³
 The modulation wave number
 $\gamma=0.2673$, $1/\gamma=3.741$

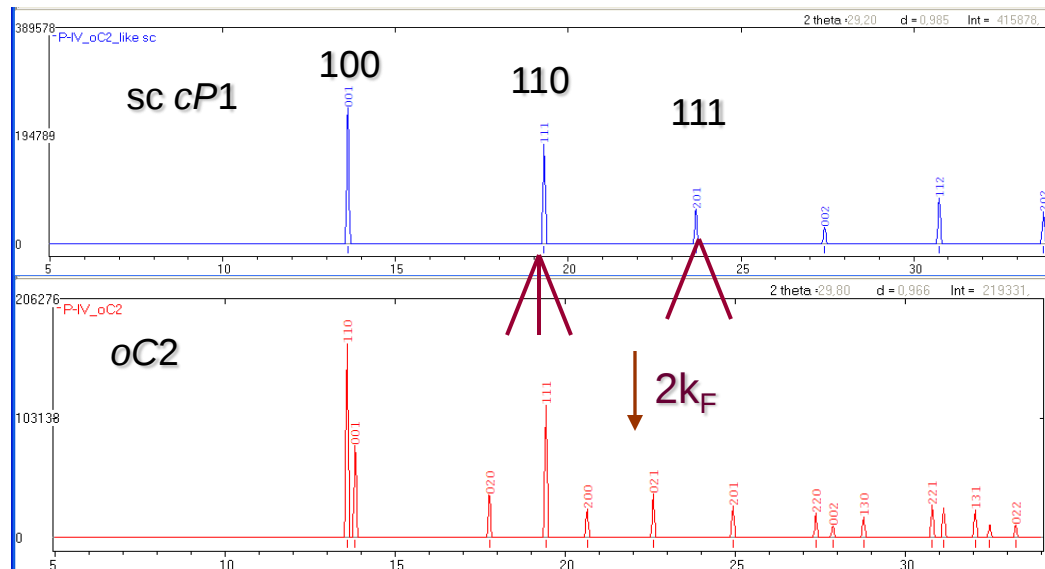
$$\mathbf{H} = h\mathbf{a}^* + k\mathbf{b}^* + l\mathbf{c}^* + m\gamma\mathbf{c}^*$$

Commensurate approximant
 $\gamma = 3/11$
 $(\gamma = 0.2727)$

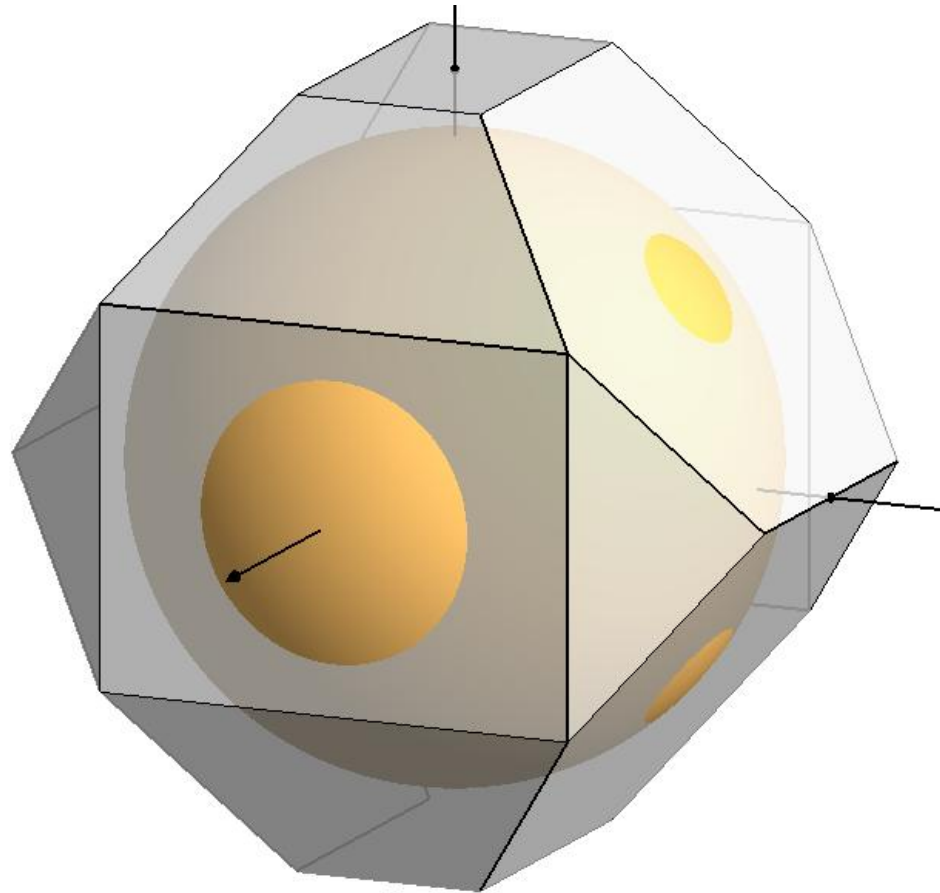
Basic cell of P-IV oC2: relation to simple cubic



$$\begin{aligned} c_{\text{ort}} &\approx a_{\text{cub}} \\ a_{\text{ort}} &\approx a_{\text{cub}} \sqrt{2} \\ b_{\text{ort}} &\approx a_{\text{cub}} \sqrt{2} \end{aligned}$$



The oC2 P-IV structure: 5 valence electrons



Main reflections:

200

021

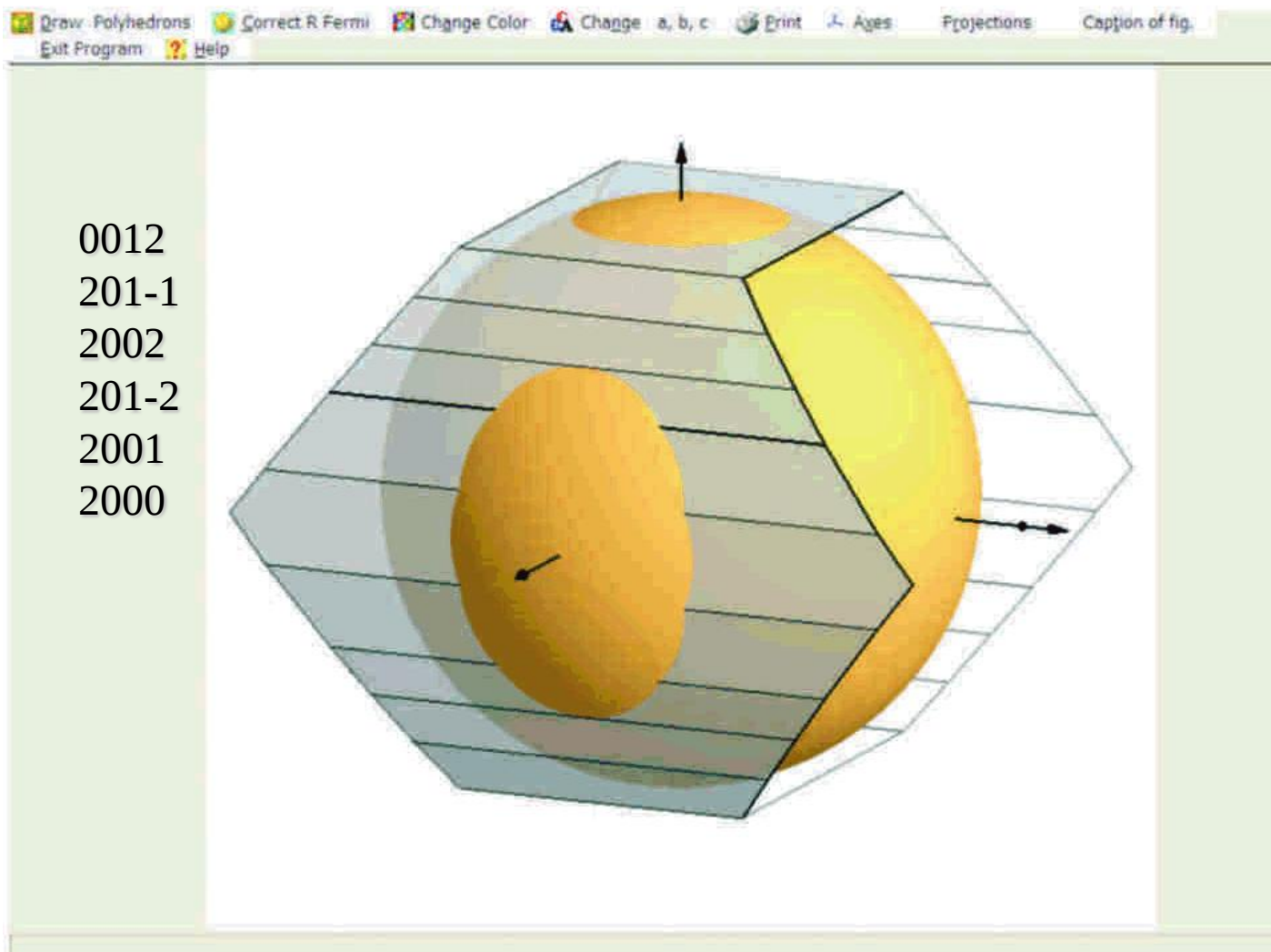
220

002

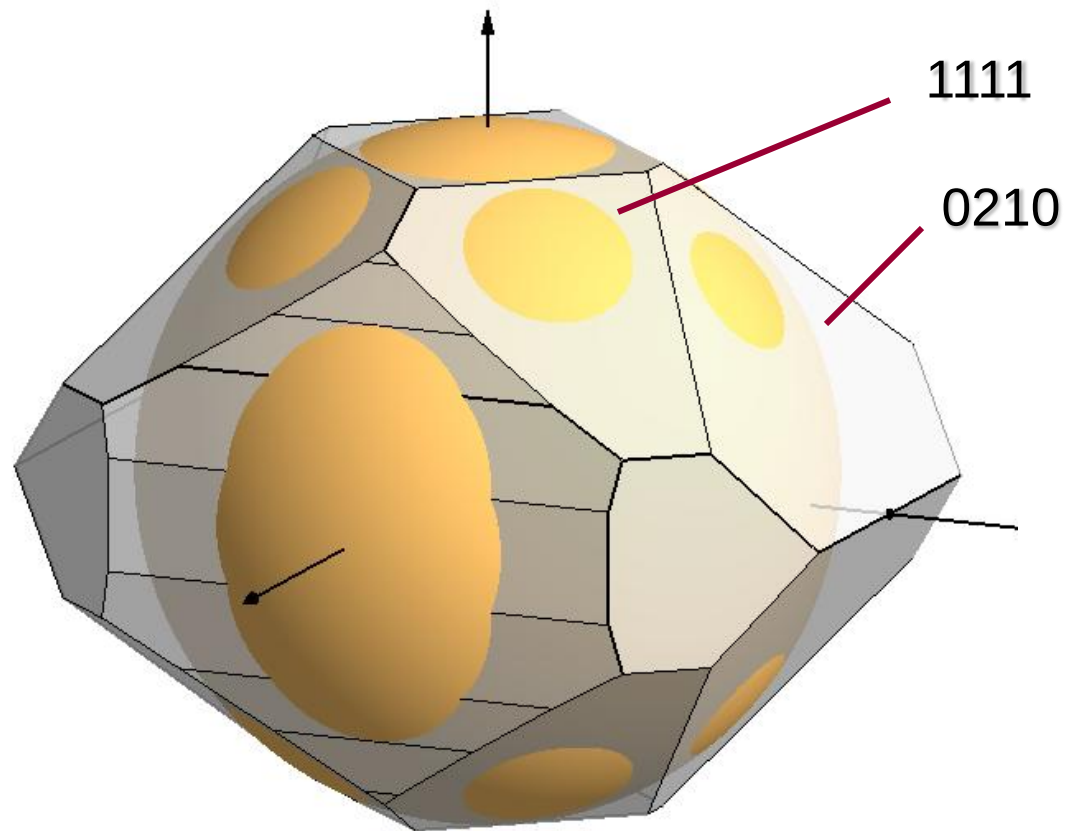
201

$$V(\text{FS}) / V(\text{BZ}) = 69\%$$

The oC2 P-IV structure: 5 valence electrons



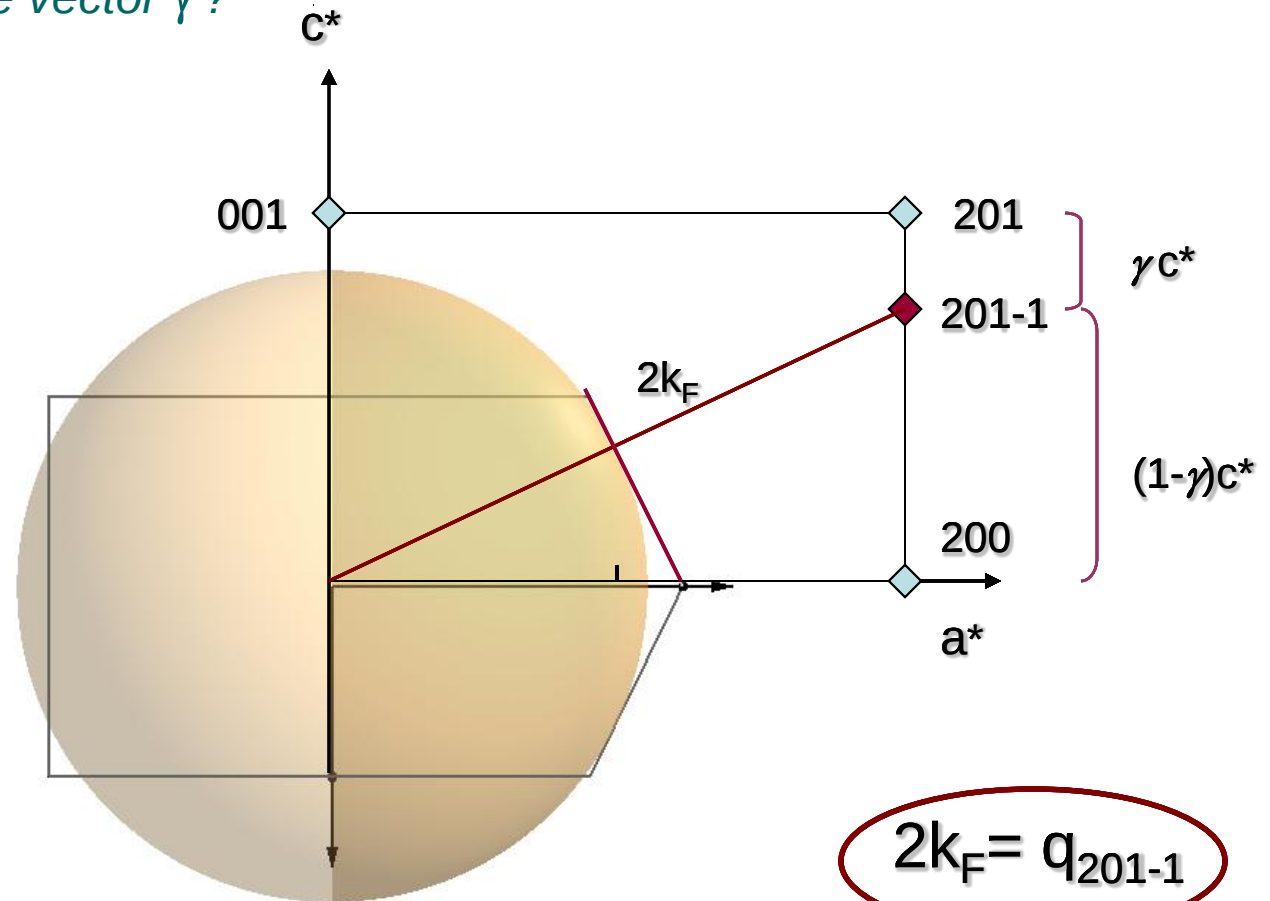
The oC2 P-IV structure: 5 valence electrons



$$V(\text{FS}) / V(\text{BZ}) = 85\%$$

The oC2 P-IV structure: 5 valence electrons

How we can define wave vector γ ?



$$(2k_F)^2 = (2a^*)^2 + ((1 - \gamma) c^*)^2$$

$$2k_F = q_{201-1}$$

$$2k_F = 5.05 \text{ \AA}^{-1}$$

$$\gamma = 0.268$$

Calcium

[Olijnyk & Holzapfel, Phys.Lett A **100**, p 191 (1984)]

20 32 113
fcc → *bcc* → *simple cubic* → *complex (?)* < 120 GPa

$$\eta_{\text{bcc}}=0.68 \quad \eta_{\text{sc}}=0.52 \quad \Delta V(\text{bcc-sc})= - 8\% \quad \text{at } 32 \text{ GPa}$$

$$R_{\text{bcc}} / R_{\text{sc}} = (\eta_{\text{bcc}} / 0.92\eta_{\text{sc}})^{1/3} = 1.12$$

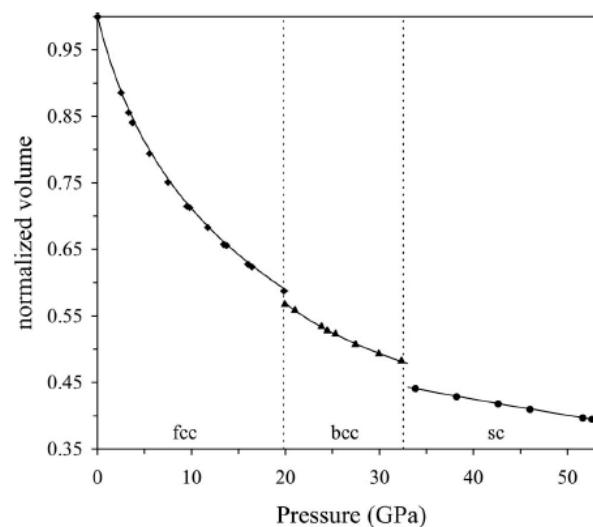
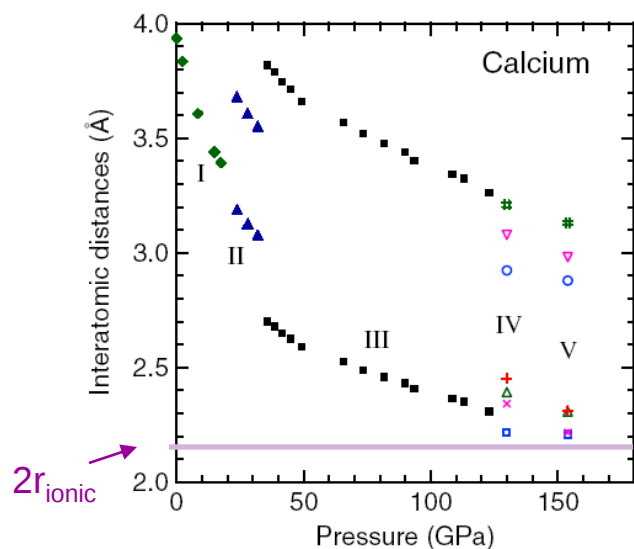
$$\Delta R(\text{bcc-sc})= -12\%$$

core overlap !?

Decrease of atomic radius
at the bcc – sc transition

Structural sequence under pressure in Ca (group-II element)

Ca	$\text{fcc} \xrightarrow{20} \text{bcc} \xrightarrow{32} \text{sc} \xrightarrow{113} tP8 \xrightarrow{139} oC8 \xrightarrow{158} tP4 < 172 \text{ GPa}$						
	10K oC2						

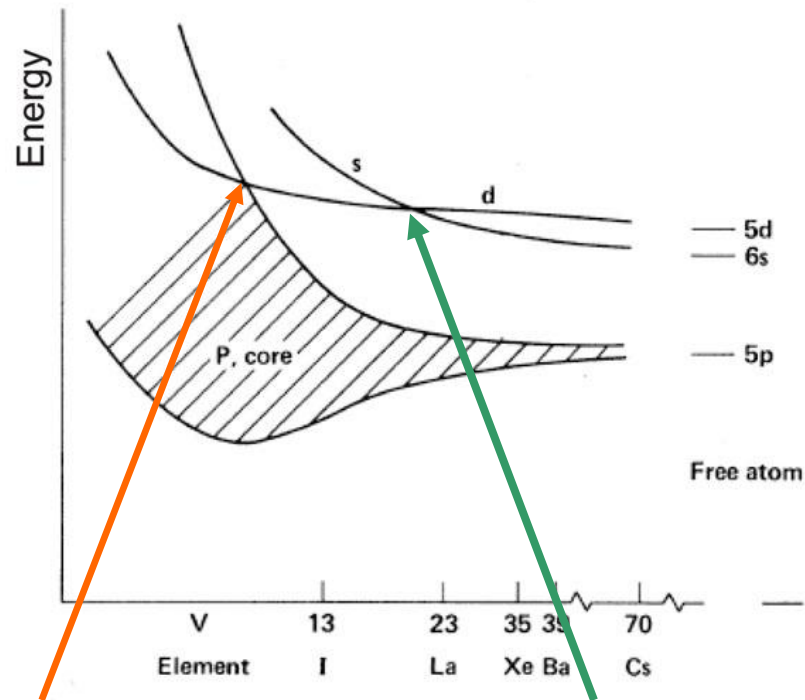


[Fujihisa *et al* PRL 2007]
 [Marques *et al* PRB 2008]
 [Nakamoto *et al* PRB 2010]
 [W Mao *et al* PNAS 2010]

Electronic energy levels

vs atomic volume

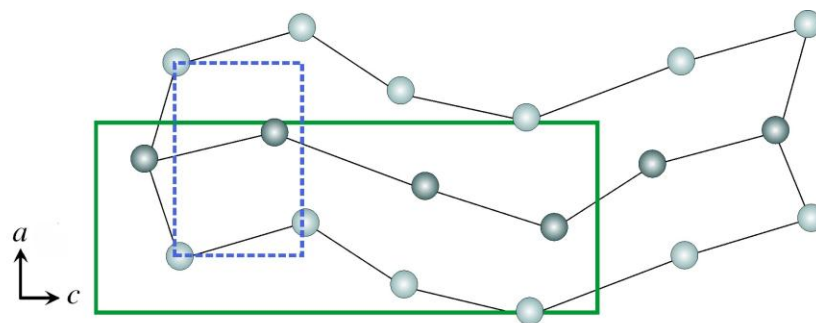
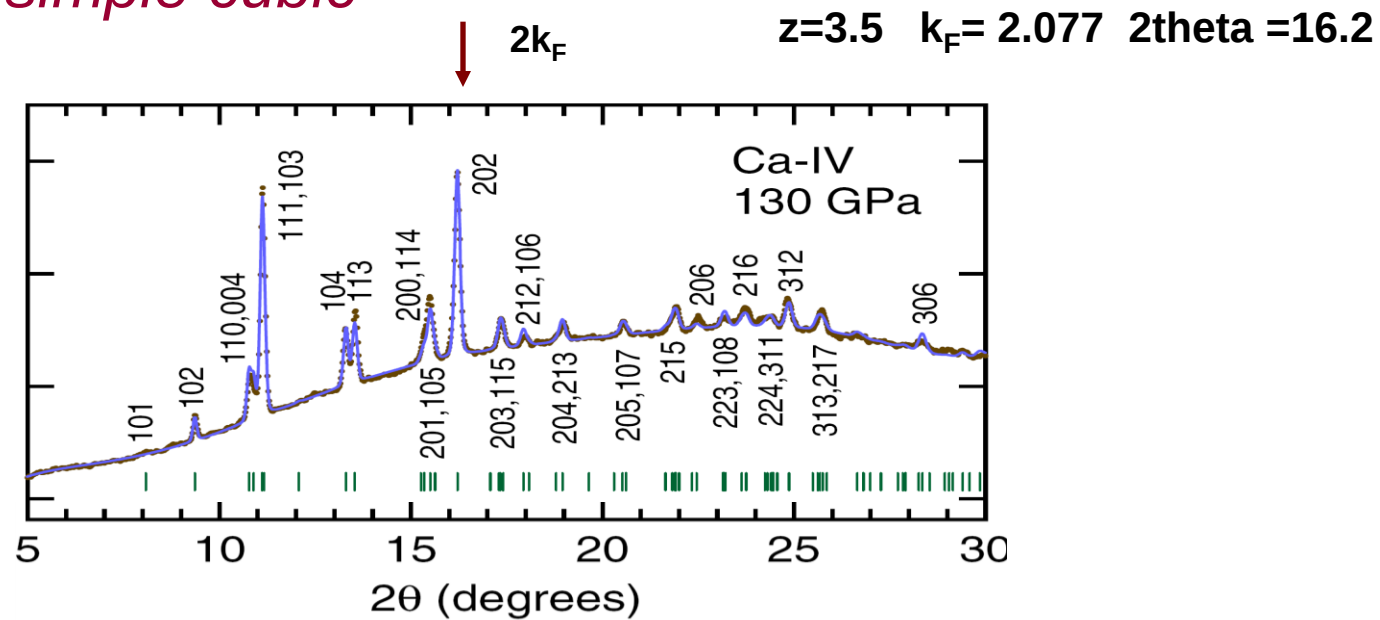
Ross & McMahan, *Phys. Rev. B* **26**,
4088 (1982)



*s-d-p(core)
hybridization*

s-d transfer

Ca post - simple cubic



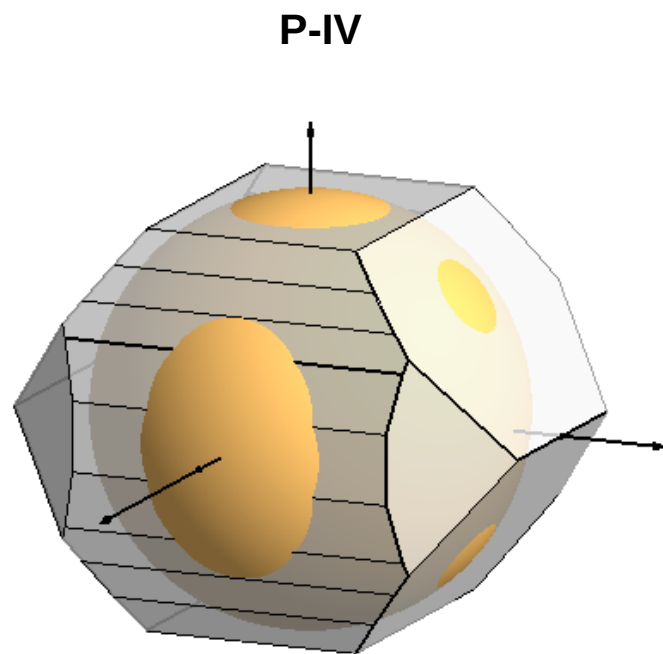
Fujihisa PRL 2008

$P=130\text{GPa}$ Ca-IV $tP8$ $a=3.209$ $c=8.97\text{ \AA}$ $(x,y,z)=(0.175; 0.527; 0.096)$

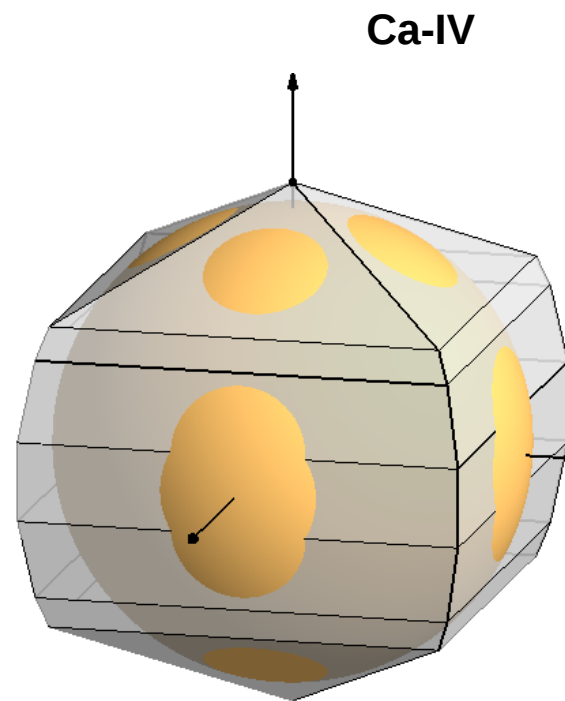
SG $P4_32_12$, 96 Atomic volume $11.553\text{ \AA}^3$

Corresponding basic cell $tP2$ $a_{\text{basic}}=3.209$ $c_{\text{basic}}=2.244\text{ \AA}$
 $=c/4$

Common features of BZ formation:



FS $z=5$
h0lm planes



FS $z=3.5$
h0l planes

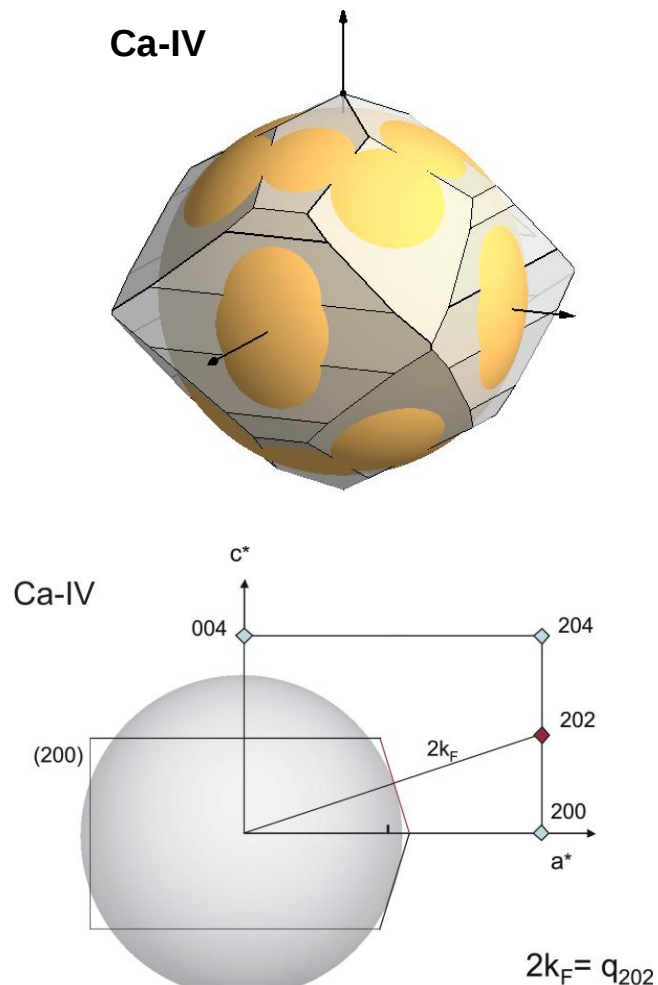
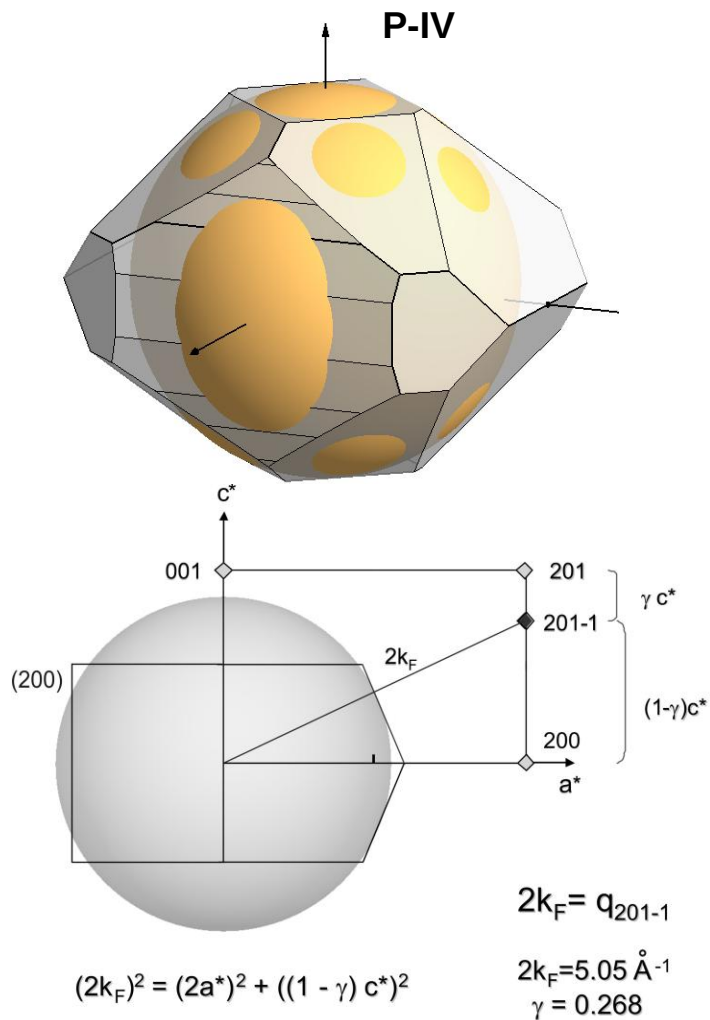
200
201
202
105

Common features of BZ formation:

In addition to 200 are formed planes

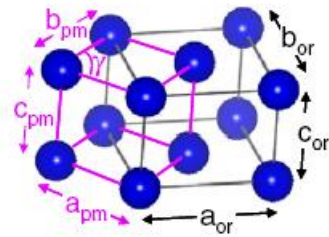
201-1 for P-IV or **201-2 (=2002)** for Ca-IV

For Ca-IV tP8 basic cell is tP2 (c/4)

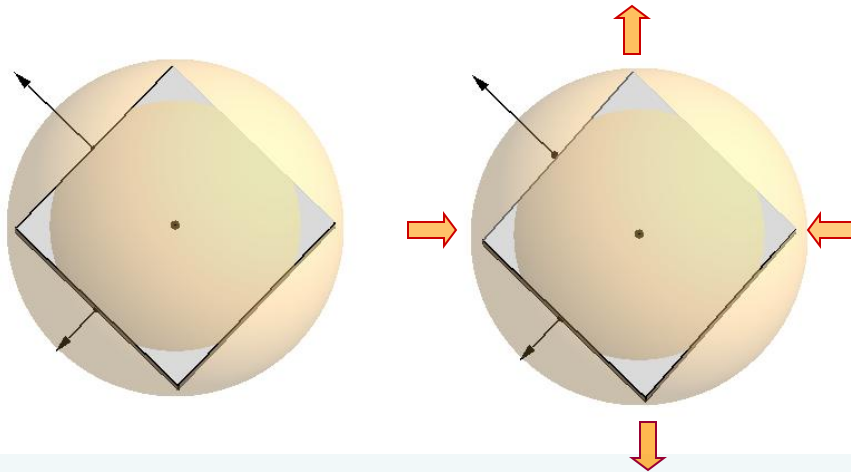


Low temperature distortion of sc-Ca

orthorhombic,
C-face-centered unit cell
of $Cmmm$ space group with
 $a = 3.6513 \text{ \AA}$,
 $b = 3.8064 \text{ \AA}$,
 $c = 2.6350 \text{ \AA}$



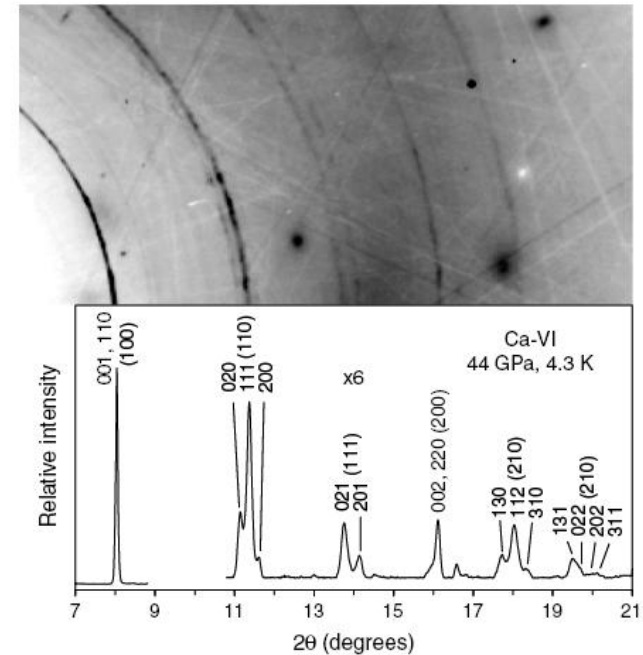
Ca-VI
Orthorhombic (or), $Cmmm$
Primitive Monoclinic (pm)



Ca sc

LT Ca-oC2

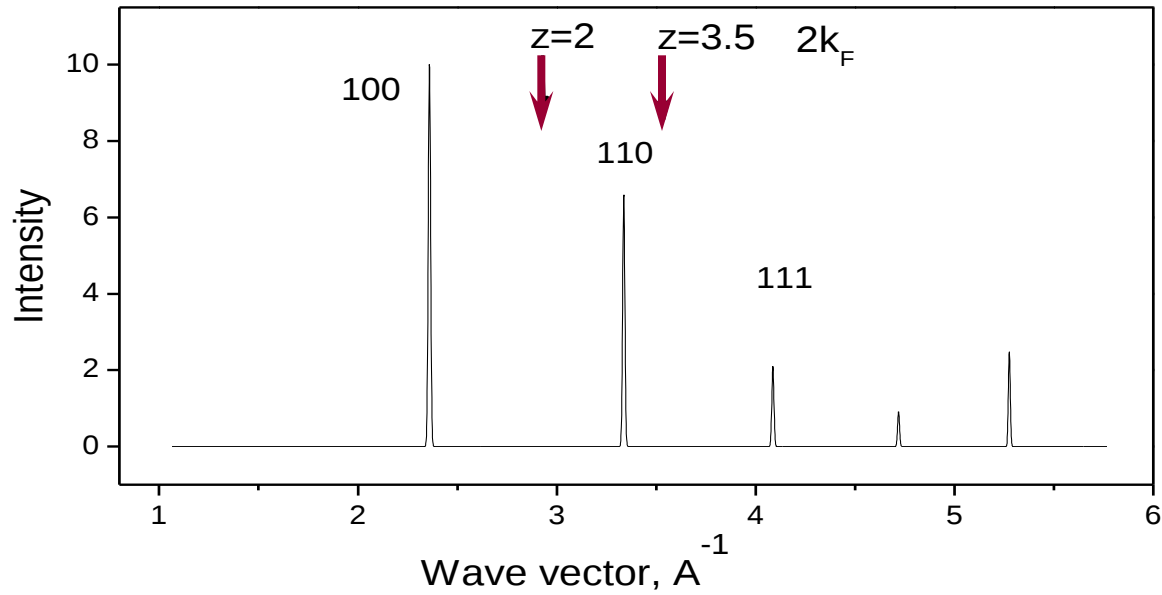
First BZ for sc and LT Ca-oC2:
projection along c^* . FS for $z=3.5$.
*Square of sc is distorted with elongation
of one diagonal and contraction of another
to contact with FS at least in one direction –
minimization of the band structure energy*



Integrated XRD pattern at 44 GPa and 4.3 K
($\lambda = 0.36940 \text{ \AA}$) collected at 16-IDB, APS.
Peaks are indexed both to the orthorhombic
structure and also to sc (in parentheses). The
small d spacing (large 2θ) region is magnified $\times 6$

[W Mao et al PNAS 2010]

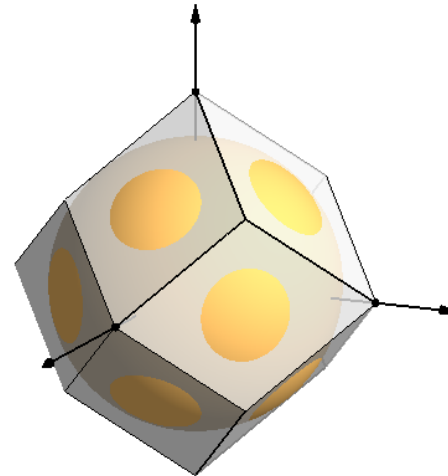
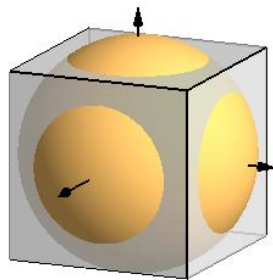
For Ca – sc how many valence electrons?



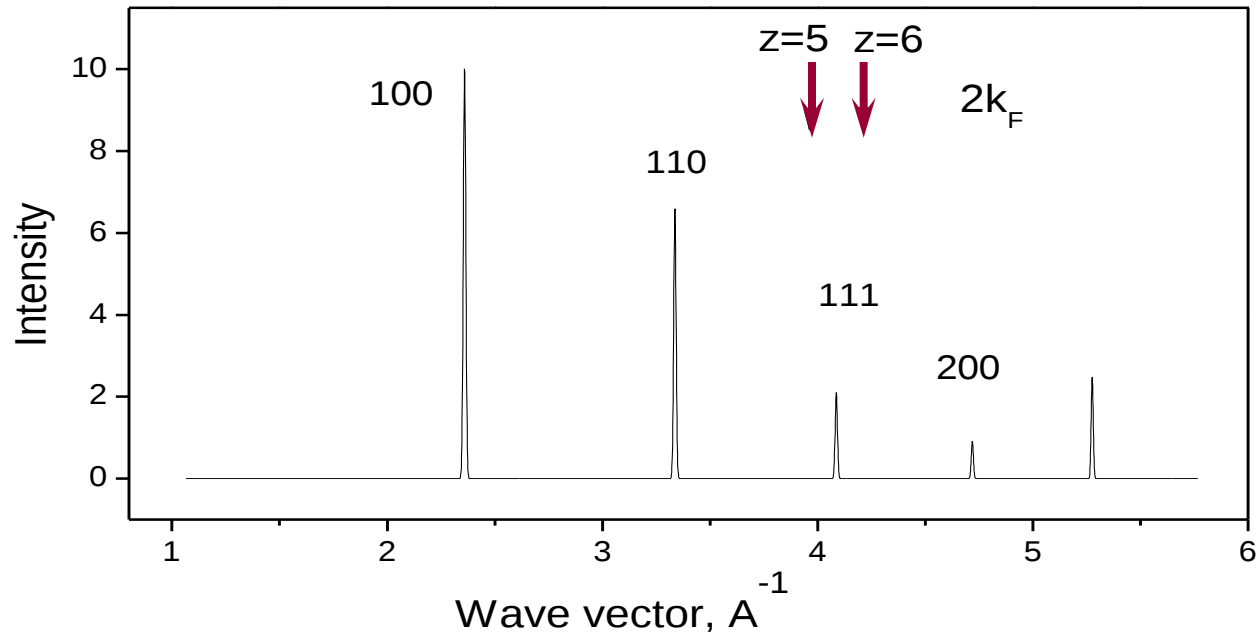
$z=2 \quad 2k_F/q_{100} = 1.24 \gg$

$z=3.5 \quad 2k_F/q_{110} = 1.056$

matching Hume-Rothery criteria !



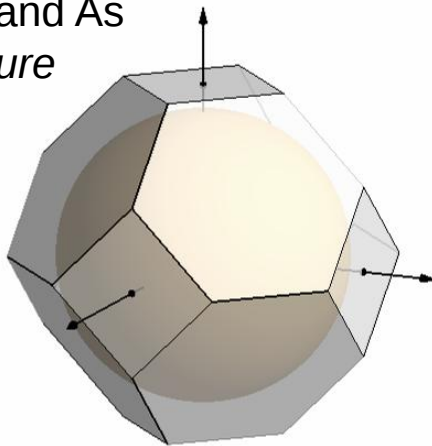
For sc how many valence electrons?



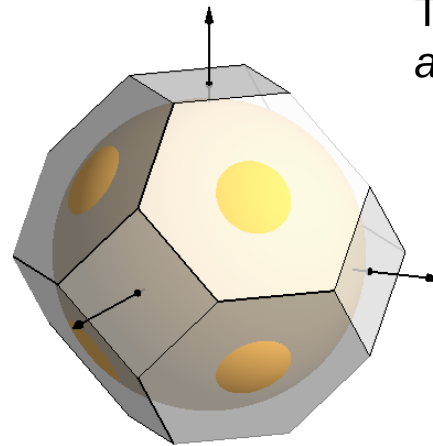
$z=5$ $2k_F/q_{111} = 0.972$

$z=6$ $2k_F/q_{111} = 1.033$

The case of sc P and As
under high pressure



The case of α -Po
ambient pressure



sc metastable phases in alloys

Constituent elements	Number of valence electrons, z	Conditions	Refs
*Au-Bi (60 at.%Bi) <i>rhombohedral</i> , 89.1° Au-Bi (70-80 at.%Bi) Au-Sb (80 at.%Sb) Au-Te (60-90 at.% Te)	*3.4 3.8 – 4.2 4.2 4.0 – 5.5	Rapid quenching from liquid	B. C. Giessen, U. Wolff and N. J. Grant, <i>Trans. met. Soc.AIME</i> , 242, 597-602 (1968)
Au-Bi (~50-100 at.% Bi) Au-Sb (68-100 at.%Sb)	~3.0 – 5 3.7 – 5	Vapor quenching and annealing	P.Haeussler et.al, Elsevier science publishers B.V, LT-17, 1984
Au-Te (60-85 at.%Te) Ag-Te (70-80 at.%Te)	4.0 – 5.25 4.5 – 5.25	Rapid quenching from liquid	ment, J. Chem. Phys. 36, 1870 (1962)
Au-Te (60-85 at.% Te)	4.0 – 5.25	Rapid quenching from liquid	C.C. Tsuei and L. R. Newkirk, <i>Phys. Rev. B</i> 18, 619–624 (1979)
Pb-Sb (50-60 at.%Sb) In-Sb (60-70 at.%Sb)	4.5 – 4.6 4.2 – 4.4	Rapid quenching from liquid	C. Borromee-Gautier, B.C. Giessen, N. J. Grant, J. Chem. Phys. 48, 1905 (1968) C.B. Jordan, J. Chem. Phys., 39, 1613 (1963)
*In-Sb (57.5 at.% Sb) <i>rhombohedral</i> , 89.2° In – Sb (60 at.% Sb)	4.15 4.2	High pressure quenching	V.F. Degtyareva, I.T. Belash, G.V. Chipenko, E.G. Ponyatovskii and V. I. Rashchupkin, Soviet Physics Solid State , 25, 1712 (1983), translated from Fiz. Tverd Tela, 25, ,2968 (1983)
Au-Sb (73-84 at.% Sb)	3.9 – 4.4	Thin film, condensed from components	L.S. Palatnik, V.M. Kosevich, L.V. Tyrina, Physics of Metals and Metallography , translated from Fizika Metallov, Metallovedenie 11, 75 (1961)
Au-Te (40-90 at.% Te)	3.0 – 5.5	Ion irradiation	J.D. Meyer and B. Stritzker, Z. Physik B 36, 47 (1979)

Electron concentration e/a

SnSb - incommensurately modulated structure

Journal of Solid State Chemistry 179 (2006) 404–412

Old friends in a new light: “SnSb” revisited

Lasse Norén^{a,*}, Ray L. Withers^a, Siegbert Schmid^b, Frank J. Brink^{a,1}, Valeska Ting^a

Inorg. Chem. 2009, 48, 5497–5503

Incommensurate Stistaite—Order Made to Order

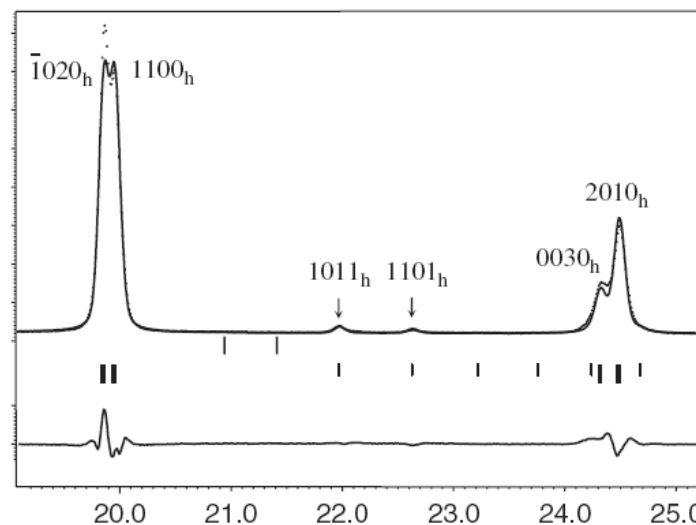
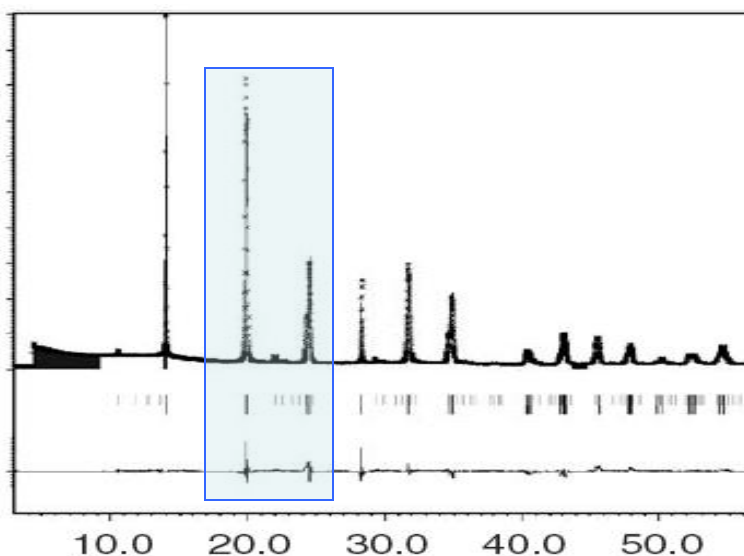
S. Lidin,^{*,†} J. Christensen,[†] K. Jansson,[†] D. Fredrickson,[†] R. Withers,[‡] L. Norén,[‡] and S. Schmid[§]

An underlying rhombohedral parent structure of space group symmetry R-3m
unit cell parameters $a = 4.3251$ Å, $c = 5.3376$ Å in the hexagonal setting

$a_R = 3.066$ Å, $\alpha_R = 89.7^\circ$ in the rhombohedral setting.

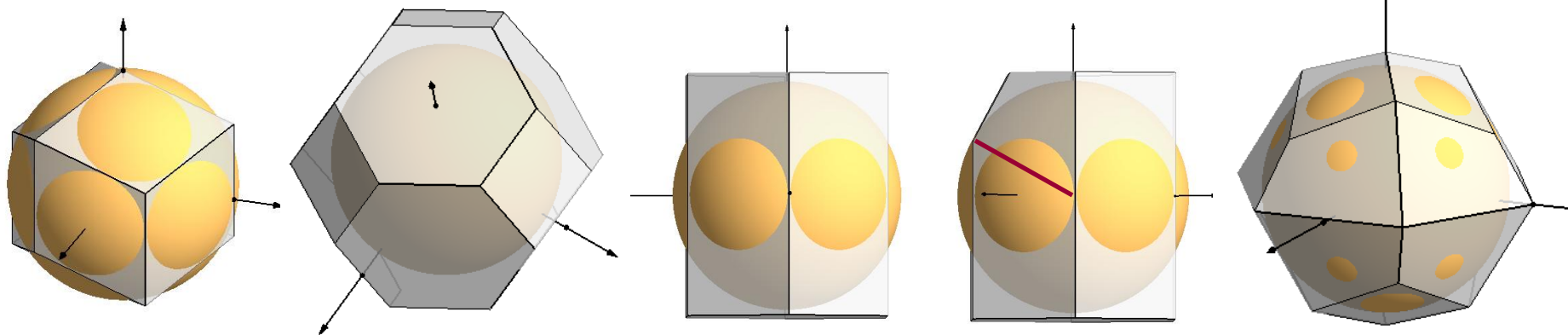
The incommensurate modulation wave vector $q=1.311c^*$

the superspace group symmetry is R-3m (0, 0, 1.311)



SnSb - incommensurately modulated structure

FS $z=4.5$



BZ planes

012 110

003 021

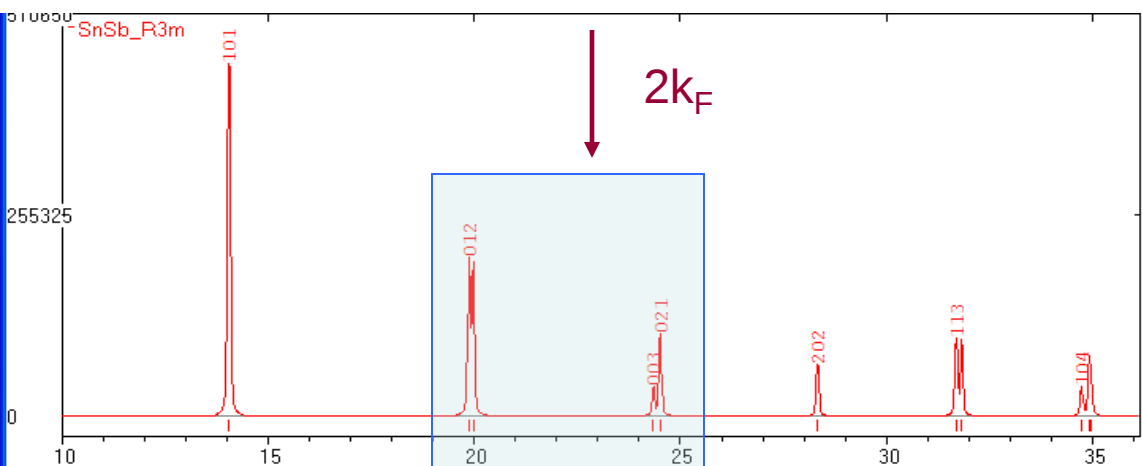
110 003 $\rightarrow$ add 1101

$$2k_F/q_{1101} = 1.01$$

1101 1011

$$2k_F/q_{1011} = 1.035$$

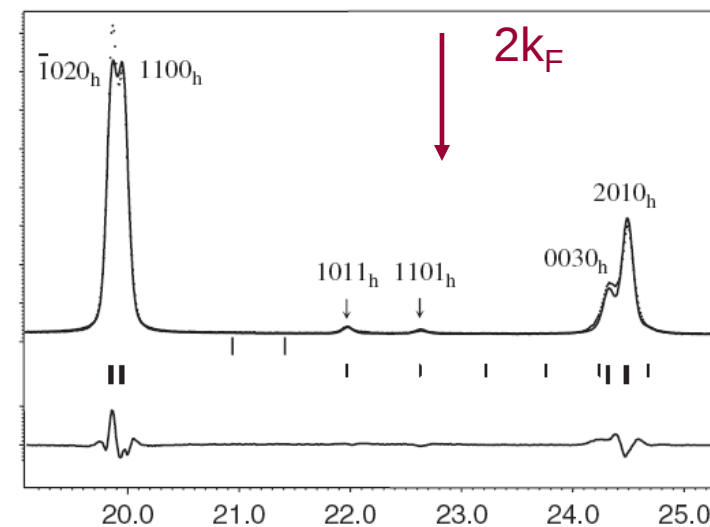
$$V_{FS}/V_{BZ} = 86\%$$



$z=4.5$

$2k_F=3.33$

$2\theta=22.95$



Aperiodic structures in elements

1. Incommensurate host-guest structures (1999-2008)
2. Incommensurately modulated structures (2003-2007)
3. Long-period superstructures (2001-2009)

IA	IIA	IIIB	IVB	VB	VIB	VII B	VIII B					IB	IIB	IIIA	IVA	VA	VIA	VIIA	VIIIA
1 H																		2 He	
3 Li	4 Be												5 B	6 C	7 N	8 O	9 F	10 Ne	
11 Na	12 Mg												13 Al	14 Si	15 P	16 S	17 Cl	18 Ar	
19 K	20 Ca	21 Sc	22 Ti	23 V	24 Cr	25 Mn	26 Fe	27 Co	28 Ni	29 Cu	30 Zn	31 Ga	32 Ge	33 As	34 Se	35 Br	36 Kr		
37 Rb	38 Sr	39 Y	40 Zr	41 Nb	42 Mo	43 Tc	44 Ru	45 Rh	46 Pd	47 Ag	48 Cd	49 In	50 Sn	51 Sb	52 Te	53 I	54 Xe		
55 Cs	56 Ba	57 La	72 Hf	73 Ta	74 W	75 Re	76 Os	77 Ir	78 Pt	79 Au	80 Hg	81 Tl	82 Pb	83 Bi	84 Po	85 At	86 Rn		
87 Fr	88 Ra	89 Ac																	
			58 Ce	59 Pr	60 Nd	61 Pm	62 Sm	63 Eu	64 Gd	65 Tb	66 Dy	67 Ho	68 Er	69 Tm	70 Yb	71 Lu			
			90 Th	91 Pa	92 U	93 Np	94 Pu	95 Am	96 Cm	97 Bk	98 Cf	99 Es	100 Fm	101 Md	102 No	103 Lr			

Conclusions

- Crystal structures of simple metals under pressure are determined by the valence electron energy term.
- Fermi sphere - Brillouin zone interactions favor the low-symmetry structures with BZ boundaries close to the Fermi sphere.
- Phosphorous-IV phase is incommensurately modulated structure due to the FS nesting effect and stabilized by the Hume-Rothery mechanism.
- Calcium high pressure phases *simple cubic* and *post – simple cubic* stabilized by increase of the number of valence electrons due to the core ionization.